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proportional to the Kelvin temperature ($P \propto T$), when the volume and number of particles is held constant. Now consider equal volumes of two gases at the same temperature (in the figures below, the sphere in the upper corner of the box is a thermometer bulb).

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AQA Chemistry Unit 4.2 - Bonding, Structure and the Properties of Matter - Higher. visit twinkl.com. The three types of chemical bonding are.... 1. ionic 2. covalent 3. metallic Describe the movement and arrangement of subatomic particles in each of the above. 1. Electrons are lost and gained to fill the outer shell.

1 AQA Chemistry Unit 4.2 - Bonding, Structure and the ...

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Chemistry Unit 4 Worksheet 2: Single Replacement Reactions Teacher initial: _____ For each reaction, use the activity series to complete the reaction. If no reaction will occur, write "no reaction". Make sure you balance your final answer. 1. $\text{Ag} + \text{KNO}_3$ _____ 2.

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Unit 4 - A-Level Chemistry

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Chemistry Unit 2 Worksheet 4: Ions 1. What is an ion? _____ 2. What does the number next to the ions signify? _____ Complete the following table, using the periodic table in the back of your book.
Element name Ion symbol Number of protons Number of electrons Number of electrons

Chemistry Unit 2 Worksheet 4: Ions Notes

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